Fission

Name -	

1.) What is the atomic number of the largest naturally occurring element on Earth?

Of the naturally occurring radioactive elements, uranium is the largest.

2.) In your own words write a definition for nuclear fission.

Nuclear fission is the splitting of a nucleus using a neutron to produce large amounts of energy.

3.) Uranium-238 is the most common isotope of uranium on the earth. Would a $10\ kg$ sample of uranium-238 be dangerous?

A $10\ kg$ sample of uranium-238 is dangerous for a few reasons. First, $10\ kg$ is over 20 pounds and could be wielded as a weapon. Secondly, and more importantly all uranium is radioactive and as such is dangerous to living organisms. As far as being dangerous as an explosive form from a nuclear fission reaction, the answer is no.

4.) Complete the following nuclear equations:

a.)
$$^{235}_{92}U + ^{1}_{0}n \rightarrow ^{139}_{56}Ba + ^{94}_{36}Kr + 3 ^{1}_{0}n$$

b.)
$$^{235}_{92}U + ^{1}_{0}n \rightarrow ^{132}_{50}Sn + ^{101}_{42}Mo + 3 ^{1}_{0}n$$

c.)
$$^{235}_{92}U + ^{1}_{0}n \rightarrow ^{101}_{41}Nb + ^{132}_{51}Sb + 3 ^{1}_{0}n$$

d.)
$$^{235}_{92}U + ^{1}_{0}n \rightarrow ^{141}_{56}Ba + ^{92}_{36}Kr + 3 ^{1}_{0}n$$

e.)
$$^{235}_{92}U + ^{1}_{0}n \rightarrow ^{90}_{37}Rb + ^{144}_{55}Cs + 2 ^{1}_{0}n$$

5.) One possible outcome of the fission reaction of uranium is the production of strontium-90 and xenon-143 along with three neutrons. Write the nuclear equation for this reaction beginning with the addition of a neutron to a uranium-235 nucleus.

The nuclear fission reaction of uranium-235 that produces xenon-143 is as follows.

$$^{235}_{92}U + ^{1}_{0}n \rightarrow ^{143}_{54}Xe + ^{90}_{38}Sr + 3 ^{1}_{0}n$$

6.) How does the total mass of the uranium-235 atom plus the neutron compare with the total mass of the products? Explain your answer.

The mass of the products of a nuclear fission reaction is less than the mass of the reactants as a small amount of mass is lost as it is converted into energy.



7.) When an atom undergoes nuclear fission, it releases a relatively large amount of energy. Where does the energy come from?

The energy of a nuclear fission reaction comes from the conversion of a small amount of mass into energy. This is described by Einstein's equation $E = mc^2$, where E is energy, m is mass, and c is the speed of light.

8.) What is meant by the term "critical mass"?

Critical mass is a term used to describe the minimum amount of radioactive material with will self sustain a fission or fusion reaction. If the mass is less than this amount then the reaction will not continue once started.