

- Level 3:**
- Sig. Figs*
- a. 3; b. 6; c. 5; d. 9; e. 2; f. 4; g. 5; h. 3; i. 4; j. 3
 - a. 90 cm; b. 72.70 m; c. 20.78 cm; d. 1.32×10^{-2} cm; e. 2.30×10^4 m; f. 73 cm; g. 9100 m; h. 0.0224 cm; i. 2.2×10^{-2} g; j. 4.5×10^5 km;
 - a. 12 cm²; b. 10 cm²; c. 3×10^7 m²; d. 3.3×10^{-8} m²; e. 0.35 mm²; f. 0.0050 mm; g. 14 km; h. 0.507 mm²; i. 56 m; j. 7.84×10^{-3} m²

- Level 3:**
- Dimen. Analysis*
- 92,100 cg; 2. 325 cm; 3. 0.30346 pt; 4. 16.0 m/s;
 5. 849 lbs/ft³; 6. 6.72×10^{20} atoms; 7. 1.7×10^{-4} L;
 - 7.5 $\times 10^{-6}$ mol; 8. 79.5 sec; 9. 6.7×10^8 mi/hr;
 10. 1.3 mol/L; 11. 8.8 mL; 12. 1.08×10^6 μ g/person

- Level 3:**
- Expo. Not.*
- a. 7.589×10^4 ; b. 10^0 ; c. 1.89×10^{-4} ; d. (no representation); e. -4.5×10^7 ; f. 8.75×10^0 ; g. -6.895×10^{-6} ; h. 1.005005×10^8 ; i. 6.09866×10^{-7} ; j. 1.25×10^{19}
 - a. 0.000,000,000,000,003907; b. 1,889,00; c. 0.0005778; d. 1.255; e. 0.000,000,0854; f. 387; g. 0.00552; h. 0.000,005555; i. 0.000,035882; j. 4,778,000,000
 - a. 2.1432×10^7 ; b. 1.2375×10^5 ; c. 3.9672×10^9 ; d. 4.0581×10^6 ; e. 7.317405×10^{17} ; f. 4.777×10^{-7} ; g. 6.528×10^{11} ; h. 1.6454×10^{-7}

- Level 3:**
- Chem. Form.*
- Pb(HCO₃)₂; 2. Mn(C₂H₃O₂)₂; 3. Na₂HPO₄; 4. HBrO₄;
 - CuSO₄ · 5H₂O; 6. P₂O₅; 7. (NH₄)₂Cr₂O₇; 8. NaCNO;
 - HClO; 10. Au₂(SO₄)₃; 11. Fe₄[Fe(CN)₆]₃; 12. KUO₄;
 - KAl(SO₄)₂; 14. N₂O; 15. Sn(MnO₄)₂; 16. H₂CrO₄;
 - NaH₂PO₄; 18. Cd(C₂H₃O₂)₂; 19. (NH₄)₂SiO₄;
 - HIO₂

- Level 3:**
- Chem. Nomen.*
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|---|--|
| 1. mercury (I) or mercurous chloride | 12. sodium dihydrogen phosphate |
| 2. dinitrogen pentoxide | 13. lead (II) or plumbous acetate |
| 3. cobalt (II) or cobaltous sulfate hexahydrate | 14. chromium (III) or chromous sulfite |
| 4. iron (II) or ferrous selenate | 15. aluminum permanganate |
| 5. erbium oxide | 16. mercury (II) or mercuric nitride |
| 6. potassium telluride | 17. potassium aluminum sulfate |
| 7. lithium hydride | 18. silicon tetrafluoride |
| 8. fluorous acid | 19. gold (II) or auric phosphide |
| 9. tin (IV) or stannic iodide or tetraiodide | 20. iron (III) or ferric ferricyanide |
| 10. arsenic (V) or arsenic oxide or pentoxide | |
| 11. osmium tetrachloride | |

Level 2:

Rx.
Predict

1. SR; yes; $\text{Sn} + \text{CuSO}_4 \rightarrow \text{Cu} + \text{SnSO}_4$
2. DR; yes; $2 \text{Fe}(\text{NO}_3)_3 + 3 \text{Na}_2\text{CrO}_4 \rightarrow \text{Fe}_2(\text{CrO}_4)_3 + 6 \text{NaNO}_3$
3. S; yes; $\text{Ca} + \text{I}_2 \rightarrow \text{CaI}_2$
4. SR; yes; $\text{Mg} + 2 \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2 \uparrow$
5. D; yes; $2 \text{CaO} \xrightarrow{\Delta} 2 \text{Ca} + \text{O}_2$
6. S; yes; $\text{C} + \text{O}_2 \xrightarrow{\Delta} \text{CO}_2 \uparrow$
7. DR; ~~yes~~^{no}; $\text{Na}_2\text{CO}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{Na}_2\text{SO}_4 + \text{H}_2\text{CO}_3 (= \text{H}_2\text{O} + \text{CO}_2 \uparrow)$
8. D; yes; $\text{FeS} \xrightarrow{\text{melt}} \text{Fe} + \text{S}$
9. SR; no; $\text{Pt} + \text{Pb}(\text{NO}_3)_2$
10. S; yes; $\text{Li}_2\text{O} + \text{H}_2\text{O} \rightarrow 2 \text{LiOH}$

Level 3:

Equation
Writing

1. $\text{N}_2 + 3 \text{H}_2 \rightarrow 2 \text{NH}_3 \uparrow$
2. $2 \text{C}_4\text{H}_{10} + 13 \text{O}_2 \rightarrow 8 \text{CO}_2 \uparrow + 10 \text{H}_2\text{O} \uparrow$
3. $2 \text{Al}_2\text{O}_3 \rightarrow 4 \text{Al} + 3 \text{O}_2 \uparrow$
4. $\text{C}_2\text{H}_5\text{OH} + 2 \text{O}_2 \rightarrow 2 \text{CO} \uparrow + 3 \text{H}_2\text{O} \uparrow$
5. $2 \text{N}_2 + 5 \text{O}_2 \rightarrow 2 \text{N}_2\text{O}_5$
6. $2 \text{C}_8\text{H}_{18} + 25 \text{O}_2 \xrightarrow{\Delta} 16 \text{CO}_2 \uparrow + 18 \text{H}_2\text{O} \uparrow$
7. $\text{Al}_2(\text{SO}_4)_3 + 2 \text{H}_3\text{PO}_4 \rightarrow 2 \text{AlPO}_4 + 3 \text{H}_2\text{SO}_4$
8. $\text{P}_2\text{O}_5 + 3 \text{H}_2\text{O} \rightarrow 2 \text{H}_3\text{PO}_4$
9. $4 \text{NH}_3 + 6 \text{NO} \rightarrow 5 \text{N}_2 \uparrow + 6 \text{H}_2\text{O} \uparrow$
10. $\text{Fe}_2\text{O}_3 + 3 \text{CO} \rightarrow 3 \text{CO}_2 \uparrow + 2 \text{Fe}$
11. $3 \text{Cu} + 8 \text{HNO}_3 \rightarrow 3 \text{Cu}(\text{NO}_3)_2 + 2 \text{NO} \uparrow + 4 \text{H}_2\text{O}$
12. $4 \text{FeS} + 7 \text{O}_2 \rightarrow 2 \text{Fe}_2\text{O}_3 + 4 \text{SO}_2 \uparrow$

Level 3:

Mole weight
& moles

1. a. 108.016; b. 249.69; c. 186.7192; d. 232.988
e. 258.20; f. 149.095; g. 378.0846; h. 162.118;
i. 252.10; j. 233.993
2. a. 164 g; b. 123 g; c. 21.4 g; d. 472.5 g; e. 242 g;
f. 384 g; g. 527 g; h. 108 g; i. 91.7 g; j. 1.45 g
3. a. 0.0155 mol; b. 0.00344 mol; c. 0.1065 mol;
d. 7.36×10^{-9} mol; e. 4.46 mol; f. 0.03604 mol;
g. 0.00275 mol; h. 0.1123 mol; i. 7.42×10^4 mol;
j. 5.90×10^{-7} mol

- Level 2:**
- % Comp*
1. 45.9% K; 16.5% N; 37.6% O
 2. 26.2% N; 7.5% H; 66.3% Cl
 3. 55.3% Sr; 44.7% Cl;
 4. 24.7% K; 34.8% Mn; 40.5% O
 5. 2.0% H; 32.7% S; 65.3% O
 6. 55.2% K; 14.6% P; 30.2% O
 7. 14.3% N; 4.1% H; 81.6% Br
 8. 80.1% Ba; 18.7% O; 1.2% H
 9. 79.9% Cu; 20.1% S;
 10. 92.3% C; 7.7% H

- Level 3:**
- Stoich.*
1. 9.39 g; 2. 8.67 L; 3. 14.5 g; 4. 7.94 g; 5. 69.2 g P_2O_5 ;
 - 26.3 g H_2O ;
 6. carbon in excess by 0.5 mol; 7. 43.97 g;
 8. magnesium hydroxide in excess by 0.343 mol; 9. no; HCN is in excess;
 10. yes; 34.6 L needed