

Energy Conservation

- 1.) The driver of a 1200 kg car travelling at $50.\frac{\text{km}}{\text{h}}$ applies the brakes so that there is a force of friction on the wheels of 1550 N , while the car travels a distance of 25 m .
 - a.) What is the original kinetic energy of the car?

 - b.) How much work is done by the force of friction from the brakes?

 - c.) How much of the car's kinetic energy is transformed to heat?

 - d.) What is the car's kinetic energy when the driver takes his foot off the brakes?

 - e.) What is the car's speed when the driver takes his foot off the brakes?

 - f.) How much work would the force of friction have had to do to stop the car?

 - g.) How much farther would the car have travelled with the drivers foot on the brake for the car to stop?

- 2.) It requires 502 KJ of heat energy to raise the temperature of 1.5 kg of water from $20.^{\circ}\text{C}$ to $100.^{\circ}\text{C}$. It takes a kettle with a power of 1500 W , 6.5 minutes to boil the water.
 - a.) How much electrical energy is transformed into heat energy by the kettle?

 - b.) What is the efficiency of the kettle?

- 3.) A $100.\text{ W}$ incandescent light bulb is only about 21% efficient. How much light energy per second does the bulb supply?

- 4.) A car of mass 1250 kg travels up a hill 250 m long and 18.0 m high. Starting with a speed of $50. \frac{\text{km}}{\text{h}}$ and ending with a speed of $70. \frac{\text{km}}{\text{h}}$. The force of friction on the car is 550 N .
- How much potential energy does the car have at the bottom of the hill?
 - What is the car's kinetic energy at the bottom of the hill?
 - What is the car's total energy at the bottom of the hill?
 - What is the potential energy of the car at the top of the hill?
 - How much does the car's total energy increase as it climbs the hill?
 - How much chemical energy does the car use?
- 5.) What is the mass of an object travelling at $20. \frac{\text{m}}{\text{s}}$ with a kinetic energy of $4000. \text{ J}$?
- 6.) How much time is required to raise the temperature of 1.50 kg of water ($c = 4180 \frac{\text{J}}{\text{kg}} \text{ } ^\circ\text{C}$) from 10.0°C to 90.0°C , using $1.50 \times 10^3 \text{ W}$ electric kettle that is 75.0% efficient?
- 7.) What is the final temperature of a 2.0 kg block of copper ($c = 430 \frac{\text{J}}{\text{kg}} \text{ } ^\circ\text{C}$) if it's original temperature was 23°C , and it absorbs $50\,000. \text{ J}$ of heat energy?
- 8.) If it takes $2.4 \times 10^5 \text{ J}$ to raise the temperature of a 6.0 kg mass up 15°C , what is it's specific heat capacity?

9.) A 15 g bullet travelling at $400. \frac{m}{s}$ hits a block of wood and penetrates it a distance of 20. cm before stopping. How much work is done in stopping the bullet?

10.) How much work must be done to stop a 1000. kg car travelling $100. \frac{km}{h}$?

11.) Tarzan is running at a top speed of $8.0 \frac{m}{s}$ and grabs a vine hanging vertically from a tall tree in the jungle. How high can he swing upwards? Does the length of the vine affect your answer?

12.) A flea should be able to jump to a height of 5.0 cm, but because of air resistance it only reaches 3.5 cm. What fraction of its energy is lost to air resistance?

13.) A 0.25 kg pine cone falls from a branch 20. m above the ground.

a.) With what speed would it hit the ground if air resistance could be ignored?

b.) If it actually hits the ground with a speed of $9.0 \frac{m}{s}$, what was the average force of air resistance on it?

Answers - 1a.) $1.2 \times 10^5 J$ b.) $3.9 \times 10^4 J$ c.) $3.9 \times 10^4 J$ d.) $8.1 \times 10^4 J$ e.) $12 \frac{m}{s}$ f.) $8.1 \times 10^4 J$ g.) $d = 52 m$							
2a.) $585\,000 J$	b.) 85.8%	3.) 21 J	4a.) zero	b.) $1.2 \times 10^5 J$	c.) $1.2 \times 10^5 J$	d.) $2.2 \times 10^5 J$	
e.) $3.36 \times 10^5 J$	f.) $4.74 \times 10^5 J$	5.) 20. kg	6.) 7 min 26 s	7.) 81°C	8.) $2.7 \times 10^3 \frac{J}{kg}$	9.) 1200 J	
10.) $3.86 \times 10^5 J$	11.) $h = 3.3 m$ No, the vine length has no impact on height.						
12.) 30. %	13a.) $-20. \frac{m}{s}$	b.) +2.0 N					