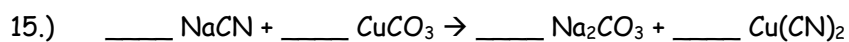
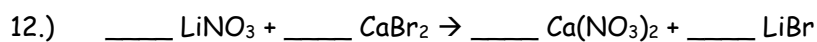
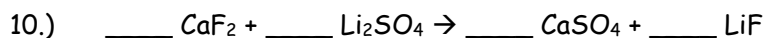
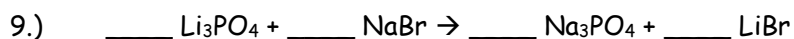
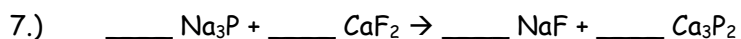
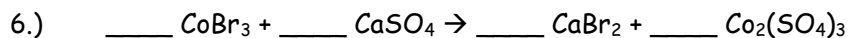
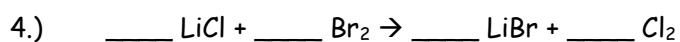


## More Balancing Equations

Name - \_\_\_\_\_

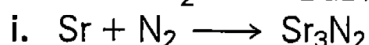
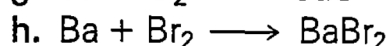
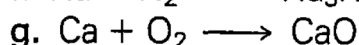
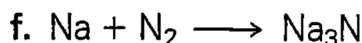
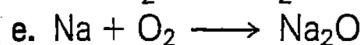
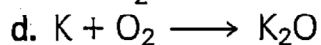
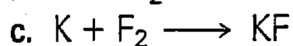
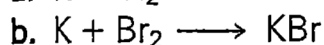
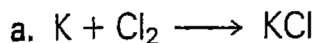
*Balance these equations!*



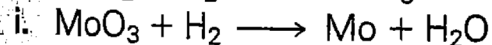
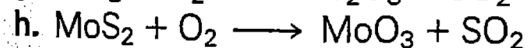
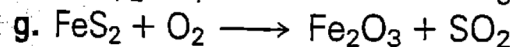
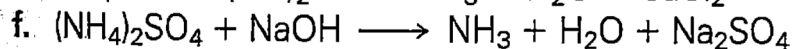
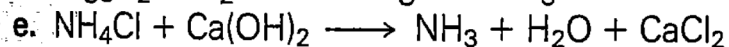
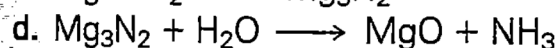
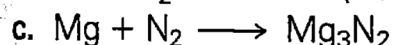
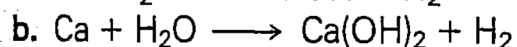
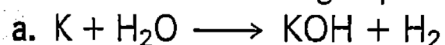
# Review and Practice

1. Answer the following for the reaction of solid carbon with oxygen gas to produce carbon dioxide gas:
  - a. Name the reactants.
  - b. List some macroscopic properties of the reactants and products.
  - c. Draw models that represent the reaction of the atoms (use Figure 5-7 as a guide).
  - d. Write the equation for the reaction.
  - e. Balance the equation.

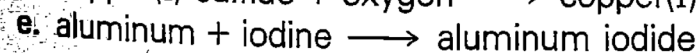
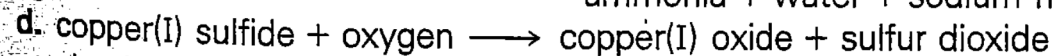
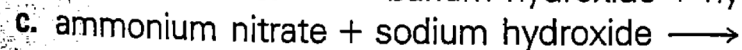
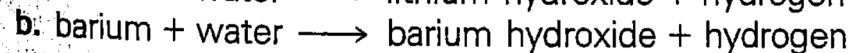
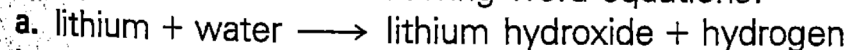
2. Balance the following equations:



3. Balance the following equations:



4. Write and balance the following word equations:



5. Balance the following equations:

