## Lewis Dot Diagram Drawings

Name - $\qquad$
Indicate the \# of VALENCE electrons for each species. Write the correct Lewis electron-dot structure for each.

| F <br> \# of electrons = $\qquad$ | $0$ <br> of electrons = | K <br> \# of electrons = $\qquad$ | Al <br> \# of electrons = $\qquad$ |
| :---: | :---: | :---: | :---: |
| $F^{-1}$ <br> \# of electrons = $\qquad$ | $0^{-2}$ <br> \# of electrons = $\qquad$ | $\mathrm{K}^{+1}$ <br> \# of electrons = $\qquad$ | $\mathrm{Al}^{+3}$ <br> \# of electrons = $\qquad$ |
| $F_{2}$ <br> \# of electrons = $\qquad$ | $\mathrm{H}_{2}$ <br> \# of electrons = | $\mathrm{HF}$ <br> \# of electrons = $\qquad$ | $\mathrm{NH}_{3}$ <br> \# of electrons = $\qquad$ |
| $\mathrm{CH}_{4}$ <br> \# of electrons = $\qquad$ | $\mathrm{NF}_{3}$ <br> \# of electrons = $\qquad$ | $\mathrm{SiF}_{4}$ <br> \# of electrons = $\qquad$ | $\mathrm{C}_{2} \mathrm{H}_{6}$ <br> \# of electrons = $\qquad$ |
| $\mathrm{MgH}_{2}$ <br> \# of electrons = | LiF <br> \# of electrons = | $\begin{aligned} & \mathrm{BaCl}_{2} \\ & \# \text { of electrons = } \end{aligned}$ | $\mathrm{FeCl}_{3}$ <br> \# of electrons = $\qquad$ |

