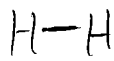


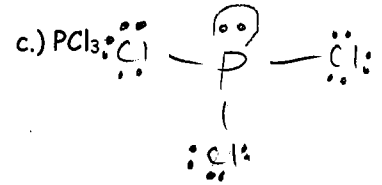
More Lewis Diagram Drawings Practice

1.) Write Lewis structures for the following: (please note, none of the solutions are using the expanded octet rule or formal charges).

a.) H₂



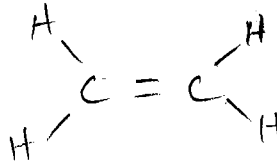
b.) HBr



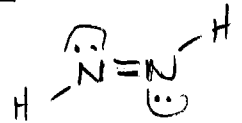
d.) SF₂



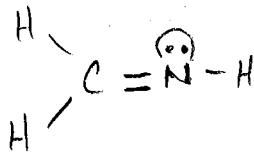
e.) H₂CCH₂



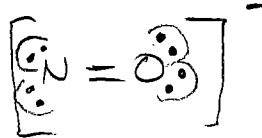
f.) HNNH



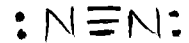
g.) H₂CNH



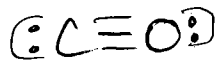
h.) NO⁻



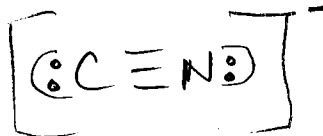
i.) N₂



j.) CO

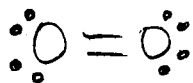


k.) CN⁻

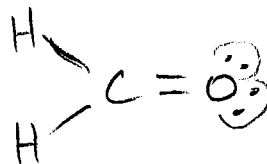


2.) Write Lewis structures for the following: (please note, none of the solutions are using the expanded octet rule or formal charges).

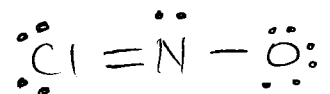
a.) O₂



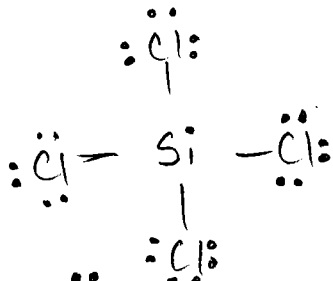
b.) H₂CO



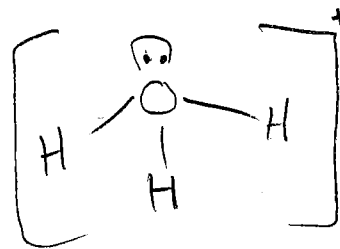
c.) ClNO



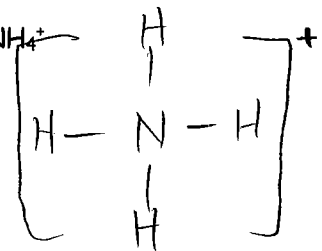
d.) SiCl₄



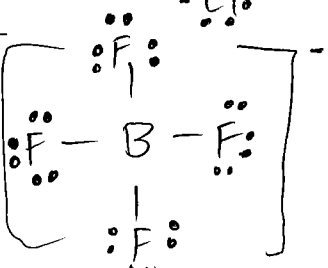
e.) H₃O⁺



f.) NH₄⁺



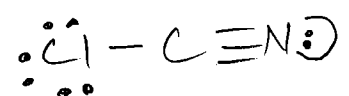
g.) BF₄⁻



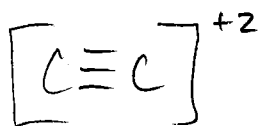
h.) HCCH



i.) ClCN

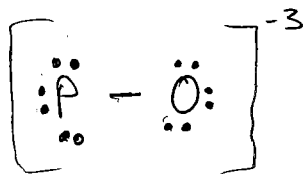


j.) C₂⁺²

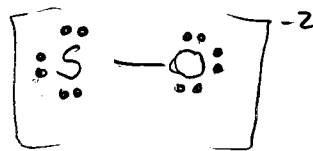


3.) Write Lewis structures for: (please note, none of the solutions are using the expanded octet rule or formal charges).

a.) PO^{3-}



b.) SO^{2-}



c.) HONO

