

<p>HCN 1+4+5=10</p> $\text{H} \cdots \text{C} \cdots \text{N}$	$\text{H} : \text{C} \equiv \text{N} :$ $\text{H} - \text{C} \equiv \text{N} :$	<p>BH₃ 3+3=6</p> $\begin{array}{c} \text{H} \\ \vdots \\ \text{H} \cdots \text{B} \cdots \text{H} \end{array}$	$\text{H} : \text{B} : \text{H}$ $\text{H} - \text{B} - \text{H}$
<p>HCSCl 1+4+6+7=18</p> $\begin{array}{c} \text{S} \\ \vdots \\ \text{H} \cdots \text{C} \cdots \text{Cl} \end{array}$	$\text{H} : \text{C} : \text{C} :$ $\text{H} - \text{C} = \text{C} :$	<p>NO₂⁻ 5+12+1=18</p> $[\text{O} \cdots \text{N} \cdots \text{O}]^-$	$:\ddot{\text{O}} : \ddot{\text{N}} : \ddot{\text{O}} :$ $[\ddot{\text{O}} - \ddot{\text{N}} = \ddot{\text{O}}]^- \text{ or } [\ddot{\text{O}} = \ddot{\text{N}} - \ddot{\text{O}}]^-$
<p>NO⁺ 5+6-1=10</p> $[\text{N} \cdots \text{O}]^+$	$:\text{N} :: \text{O} :$ $[:\text{N} \equiv \text{O}:]^+$	<p>NCOH 5+4+6+1=16</p> $\text{H} \cdots \text{O} \cdots \text{N} \cdots \text{C}$	$\text{H} : \ddot{\text{O}} : \text{N} : \ddot{\text{C}} :$ $\text{H} - \ddot{\text{O}} = \text{N} = \ddot{\text{C}}$ $\text{H} - \text{O} \equiv \text{N} - \ddot{\text{C}} \quad \text{H} - \ddot{\text{O}} - \text{N} \equiv \text{C} :$
<p>NH₂⁻</p> $[\text{H} \cdots \text{N} \cdots \text{H}]^-$	$[\text{H} - \ddot{\text{N}} - \text{H}]^-$	<p>SO₂</p> $\text{O} \cdots \text{S} \cdots \text{O}$	$\ddot{\text{O}} = \ddot{\text{S}} - \ddot{\text{O}} : \text{ or } :\ddot{\text{O}} - \ddot{\text{S}} = \ddot{\text{O}}$
<p>S₂Cl₂</p> $\text{Cl} \cdots \text{S} \cdots \text{S} \cdots \text{Cl}$	$:\ddot{\text{Cl}} - \ddot{\text{S}} - \ddot{\text{S}} - \ddot{\text{Cl}} :$	<p>N₂H₄</p> $\begin{array}{c} \text{H} \quad \text{H} \\ \vdots \quad \vdots \\ \text{N} \cdots \text{N} \\ \vdots \quad \vdots \\ \text{H} \quad \text{H} \end{array}$	$\begin{array}{c} \text{H} \quad \text{H} \\ \vdots \quad \vdots \\ \text{N} - \text{N} \\ \vdots \quad \vdots \\ \text{H} \quad \text{H} \end{array}$
<p>SF₆</p> $\begin{array}{c} \text{F} \\ \vdots \\ \text{F} \cdots \text{S} \cdots \text{F} \\ \vdots \\ \text{F} \end{array}$	$\begin{array}{c} \text{F} \\ \vdots \\ \text{F} - \text{S} - \text{F} \\ \vdots \\ \text{F} \end{array}$	<p>CNO⁻</p> $[\text{C} \cdots \text{N} \cdots \text{O}]^-$	$[\ddot{\text{C}} = \ddot{\text{N}} = \ddot{\text{O}}]^- \text{ or } [\ddot{\text{C}} - \ddot{\text{N}} = \ddot{\text{O}}]^-$ $[:\text{C} \equiv \text{N} - \ddot{\text{O}}:]^-$
<p>NO₃⁻</p> $[\text{O} \cdots \text{N} \cdots \text{O}]^-$	$[\ddot{\text{O}} = \ddot{\text{N}} - \ddot{\text{O}}]^- \quad [\ddot{\text{O}} - \ddot{\text{N}} = \ddot{\text{O}}]^- \quad [\ddot{\text{O}} - \ddot{\text{N}} = \ddot{\text{O}}]^-$	<p>C₄H₄</p> $\begin{array}{c} \text{H} \cdots \text{C} \cdots \text{C} \cdots \text{H} \\ \vdots \quad \vdots \\ \text{H} \cdots \text{C} \cdots \text{C} \cdots \text{H} \end{array}$	$\text{H} - \text{C} = \text{C} - \text{H}$ $\text{H} - \text{C} = \text{C} - \text{H} \quad \text{or} \quad \text{H} - \text{C} - \text{C} - \text{H}$
<p>N₂O₃</p> $\text{O} \cdots \text{N} \cdots \text{N} \cdots \text{O}$	$\ddot{\text{O}} = \ddot{\text{N}} - \ddot{\text{N}} = \ddot{\text{O}} \quad \text{or} \quad \ddot{\text{O}} - \ddot{\text{N}} - \ddot{\text{N}} = \ddot{\text{O}}$ $\ddot{\text{O}} - \ddot{\text{N}} = \ddot{\text{N}} = \ddot{\text{O}} \quad \ddot{\text{O}} - \ddot{\text{N}} - \ddot{\text{N}} = \ddot{\text{O}}$	<p>CH₃CO₂H</p> $\begin{array}{c} \text{H} \quad \text{O} \\ \vdots \quad \vdots \\ \text{H} \cdots \text{C} \cdots \text{C} \cdots \text{O} \cdots \text{H} \\ \vdots \\ \text{H} \end{array}$	$\text{H} - \text{C} - \text{C} - \ddot{\text{O}} - \text{H}$
<p>SeBr₄</p> $\begin{array}{c} \text{Br} \\ \vdots \\ \text{Br} \cdots \text{Se} \cdots \text{Br} \\ \vdots \\ \text{Br} \end{array}$	$\begin{array}{c} \text{Br} \\ \vdots \\ \text{Br} - \text{Se} - \text{Br} \\ \vdots \\ \text{Br} \end{array}$	$\text{H} - \text{C} - \text{C} = \ddot{\text{O}} - \text{H}$	